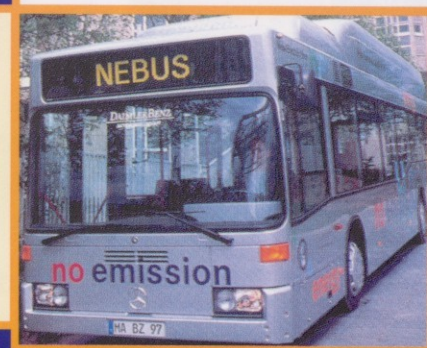
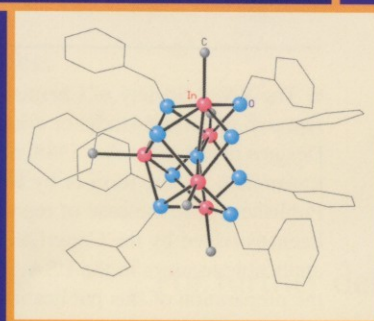
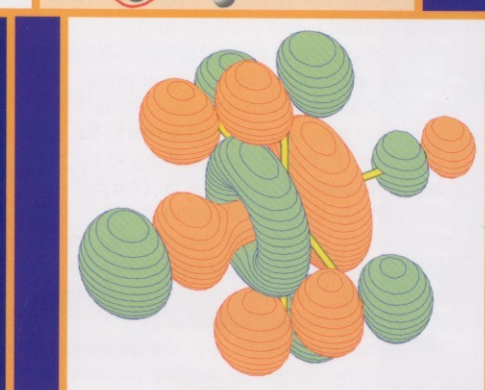
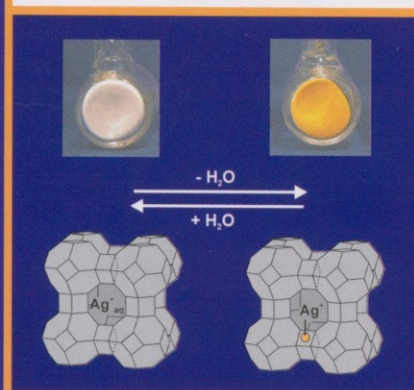
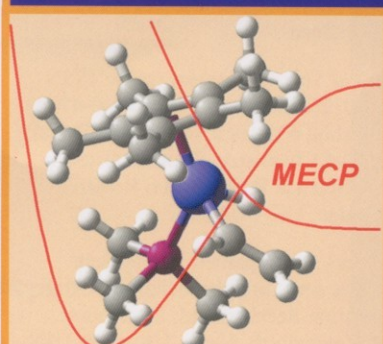
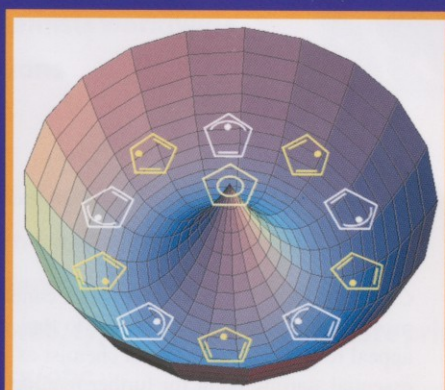


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# The electronic structure of Cu<sup>+</sup>, Ag<sup>+</sup>, and Au<sup>+</sup> zeolites



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A variety of procedures have been used to prepare d<sup>10</sup>-zeolite materials. The electronic structure of these materials can be regarded to a first approximation as a superposition of the framework, of the charge compensating ions, of solvent molecules and of guest species. Zeolite oxygen to d<sup>10</sup>-ion charge transfer transitions dominate the electronic spectra if the ions coordinate to the zeolite oxygens. Specific

coordination sites can influence the energy and the intensity of these transitions remarkably. Intra guest transitions dominate in quantum dot materials, as discussed in detail for luminescent Ag<sub>2</sub>S zeolite A. The zeolite is not needed for the photocatalytic water oxidation on Ag<sup>+</sup>/AgCl photoanodes with visible light. It can, however, be used to increase the active surface area substantially.

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*David Schürch (right) studied chemistry at the University of Calgary and the University of Berne where he obtained his first degree. In 2002 he received his PhD degree at the University of Berne.*

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